PRAB TECH BRIEF

Eliminating Soluble Heavy Metal Ions From Industrial Wastewater by Precipitation and Ultrafiltration

By Tim Hanna



Introduction

Today's manufacturers should be keenly aware of the problems—and especially the penalties—associated with discharging industrial wastewater into sewage treatment systems. The removal of soluble heavy metal ions from wastewater is a common industrial treatment requirement in modern manufacturing. Some of the processes that generate waste streams containing heavy metals are:

- Electroplating
- Electroless nickel plating
- Printed circuit board manufacturing
- Metal forming operations
- Battery recycling
- Mining operations

Precipitation and ultrafiltration provide two ways to remove heavy metals from wastewater that are superior to conventional solutions, and both can help manufacturers avoid costly consequences. In this tech brief, we will discuss how we determine which process will deliver the best results based on a variety of application-specific criteria.

We're here to make a difference.

5801 East N Avenue, Kalamazoo, MI 49048 1-800-968-7722 l sales@prab.com Several problems can occur when industrial wastewater is discharged into sewage treatment systems:

- Causing corrosion and other interference within those treatment systems
- Exposing plant personnel to toxic substances and hazardous fumes
- Incurring expensive sludge disposal costs
- Toxic pollutants passing through the plant and into surface waters

All of these can be controlled by partnering with PRAB to implement a precipitation or ultrafiltration pre-treatment process in your operation.



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REMOVING HEAVY METALS FROM WASTEWATER

Understanding the Variables

The concentration and chemical form of the soluble heavy metals in a particular wastewater stream varies, depending on the industry and the mix of operations at a processing site. For example: In the plating industry, specialty chemical suppliers have developed alloy technologies such as zinc/nickel and other alloy coatings to maximize the high corrosion resistance of the metal.

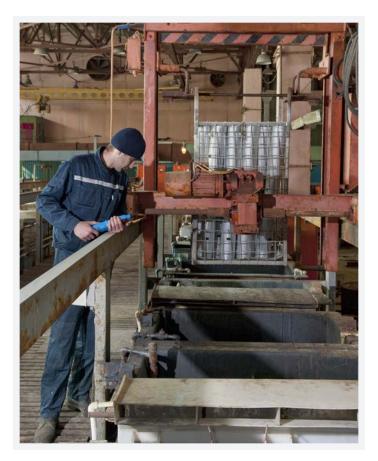
The zinc/nickel of choice for today's surface finishing industry is an alkaline electrolyte chemistry instead of an acid electrolyte chemistry. This is due to the alloy distribution required by the plate thickness throughout the entire part. Unfortunately, alkaline chemistry contains chelators and/or complexors and organic additives to allow for the co-deposition of the nickel in the deposit with the zinc. These chelators/complexors make the wastewater extremely difficult to treat through conventional chemical or physical treatment methods.

Zinc and chrome wastewater can be treated through conventional methods, such as precipitation and the use of polymers in a lamella clarifier, followed by a filter press to dewater the sludge. The highly complexed alkaline zinc/nickel, however, doesn't want to precipitate and the chelated metals tend to stay in solution. Additionally, the metal concentrations are too high to discharge to the local POTW (Publicly Owned Treatment Works). Nickel is a hazardous constituent and is not exempted. The addition of wastewater from the zinc/nickel process will cause your filter press cake to become a "listed" RCRA hazardous waste with the F006 designation, resulting in significant increases to your disposal and regulatory costs.

Removing the Solids

Typical removal strategies involve precipitating the metals in an insoluble form—such as hydroxides, sulfides, carbonates or some combination—and removing the precipitate with tubular ultrafiltration (for very high-quality filtrate) or conventional clarification. The resultant sludge is then collected, thickened and dewatered for landfill disposal and listed as F006 (but at a much smaller volume) and the water phase is returned to the ultrafiltration process.

Unlike conventional clarifier techniques, tubular ultrafiltration (UF) is a very effective method of removing virtually all of the precipitated metal hydroxide/sulfides/carbonates from the treated wastewater stream. The resulting high-quality permeate can be fed directly to reverse osmosis equipment

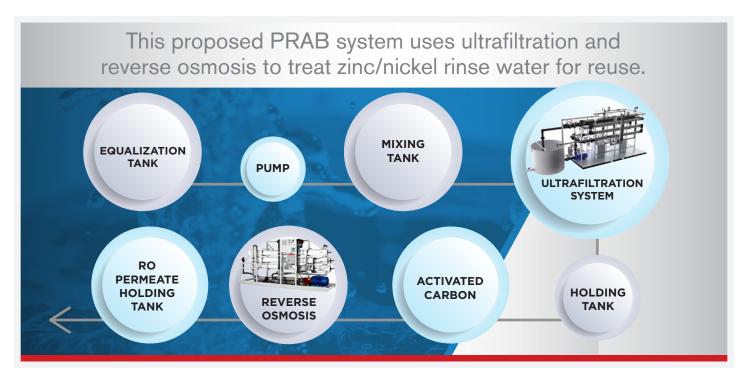


for reuse, reclaimed 'as is' or discharged to the POTW. Tubular ultrafiltration can process solids concentrations of up to 18% (w/w), but the most efficient operation normally occurs between 3% and 5% solids.

The sludge is normally drawn off and passed through conventional thickening/filter press processes or similar processes to produce dewatered solids, typically for landfill disposal. Supernatants from the sludge handling processes are recycled through the TMF system so that the only outputs are the filtered water and the compact solids cake.

Hydroxide precipitation is a common method as it is relatively simple to operate. Sulfide precipitation has some advantages, but pH and oxidation-reduction potential (ORP) must be carefully controlled to minimize the risk of producing toxic levels of hydrogen sulfide gas. Using hydroxide and sulfide precipitation in two sequential steps is also an option, particularly where complexes or chelates are present. Carbonate coprecipitation using sodium or calcium carbonate can also be helpful; for example, for soluble lead reduction, lead carbonate is essentially insoluble (0.00011 g/100 mL @ 20°C) and will precipitate out. Phosphate precipitation is another option, although this process is not as common.

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Finally, there are membrane technologies available that utilize a proprietary additive in conjunction with a chelate breaker for precipitating the chelated metals. The advantages of this process include:

- Eliminating the need for clarifiers or lamella separators
- Not requiring coagulants for processing because the pore size of the microfiltration and ultrafiltration membrane is so small
- Reducing the volume of sludge generated
- Lowering labor costs by reducing the need for operator interface and constant monitoring
- Lowering chemical costs
- Producing a more consistent effluent

Wastewater treatment can understandably appear to be a daunting task, especially when heavy metals are part of the equation. Partnering with PRAB—a proven expert in wastewater solutions for a variety of industries, chemistries and applications—can ease your burden and improve your profitability as you address an important and essential aspect of modern manufacturing.

About the Author

Tim Hanna is the Vice President of Business Development for PRAB, a designer and manufacturer of wastewater recycling systems as well as engineered conveyors and equipment for processing turnings, chips, and metalworking fluids. Since 1970, Hanna has worked closely with companies and communities around the world that take advantage of the latest fluid filtration and chip handling technologies to both lower cost and increase environmental stewardship.

About PRAB

PRAB is a leading engineer and manufacturer of conveyors and chip and fluid management systems. Its customized solutions automate metal handling, reduce labor costs, reclaim and recycle expensive cutting fluids/coolants, and maximize return on recycling metals. With its expertise honed by more than 4,500 installations for the world's leading OEMs and suppliers, PRAB continuously improves material handling, housekeeping, and compliance to environmental rules and regulations within the automotive, aerospace, medical, electronics, defense, off-road, and energy markets. For more information about PRAB, visit prab.com.

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